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# 1-(Carbamoylmethyl)-5-oxopyrrolidin-3-yl propionate

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#### **Key indicators**

Single-crystal X-ray study T = 294 KMean  $\sigma(\text{C-C}) = 0.004 \text{ Å}$  R factor = 0.049 wR factor = 0.152Data-to-parameter ratio = 12.4

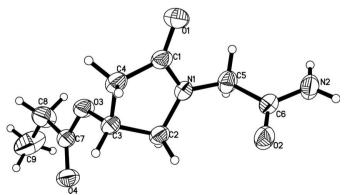
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The title compound,  $C_9H_{14}N_2O_4$ , was prepared from 2-(4-hydroxy-2-oxocyclopentyl)acetamide and propionic acid anhydride. In the crystal structure, molecules are linked through  $N-H\cdots O$  hydrogen bonds, forming an extended supramolecular assembly which contributes to the stability of the structure in the solid state.

# Comment

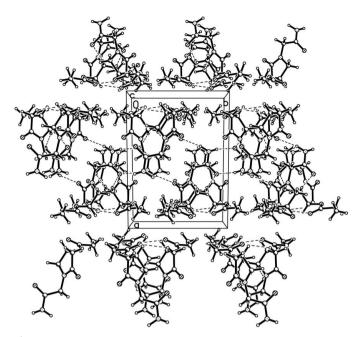
The synthesis of new intentionally designed crystal structures is part of a major strand of modern chemistry (Belloni *et al.*, 2005; Tynan *et al.*, 2005). One of the aims of crystal engineering is to establish control over the preparation of crystalline solid materials so that their subsequent architecture and properties are predictable. As part of an investigation of crystal structures with strong intermolecular bonding that might provide useful information in the field of crystal engineering, we report here the synthesis and crystal structure of the title compound, (I).

A view of the molecular structure of (I) is shown in Fig. 1 and selected bond lengths and angles in (I) are given in Table 1. They are all within normal ranges. Molecules are linked through  $N-H\cdots O$  hydrogen bonds (Table 2), forming a supramolecular assembly that extends throughout the crystal structure, leading to enhanced stability. Fig. 2 shows a portion of this extensively hydrogen-bonded supramolecular assembly.



**Figure 1** A view of the title compound, shown with 30% probability displacement ellipsoids.

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**Figure 2** A view down the *c* axis of a portion of the crystal structure, showing the extensive intermolecular hydrogen-bonding interactions (dashed lines).

## **Experimental**

An anhydrous pyridine solution of 2-(4-hydroxy-2-oxocyclopentyl)-acetamide (1.54 g, 10 mmol) was added to propionic acid anhydride (1.30 g, 10 mmol). The mixture was stirred at 298 K for 20 h under nitrogen, and a white precipitate formed. This product was isolated and recrystallized from pyridine, and then dried *in vacuo* to give the pure compound in 98% yield. White single crystals of (I) suitable for X-ray analysis were obtained by the slow evaporation of a pyridine solution of (I).

#### Crystal data

$C_9H_{14}N_2O_4$	$D_x = 1.340 \text{ Mg m}^{-3}$
$M_r = 214.22$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 102:
a = 12.6981 (17)  Å	reflections
b = 9.0630 (13)  Å	$\theta = 2.8 - 21.3^{\circ}$
c = 9.5362 (13)  Å	$\mu = 0.11 \text{ mm}^{-1}$
$\beta = 104.592 (2)^{\circ}$	T = 294 (2)  K
$V = 1062.1 (3) \text{ Å}^3$	Block, white
Z = 4	$0.32 \times 0.20 \times 0.16 \text{ mm}$

## Data collection

Bruker SMART CCD area-detector	1811 independent reflections
diffractometer	993 reflections with $I > 2\sigma(I)$
$\varphi$ and $\omega$ scans	$R_{\rm int} = 0.033$
Absorption correction: multi-scan	$\theta_{\rm max} = 25.0^{\circ}$
(SADABS; Bruker, 1999)	$h = -8 \rightarrow 15$
$T_{\min} = 0.901, T_{\max} = 0.983$	$k = -10 \rightarrow 10$
5641 measured reflections	$l = -8 \rightarrow 11$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_0^2) + (0.0912P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.049$	where $P = (F_0^2 + 2F_c^2)/3$
$wR(F^2) = 0.152$	$(\Delta/\sigma)_{\rm max} < 0.001$
S = 0.97	$\Delta \rho_{\text{max}} = 0.23 \text{ e Å}^{-3}$
1811 reflections	$\Delta \rho_{\min} = -0.20 \text{ e Å}^{-3}$
146 parameters	Extinction correction: SHELXL97
H atoms treated by a mixture of	Extinction coefficient: 0.025 (6)
independent and constrained	
refinement	

Table 1 Selected geometric parameters ( $\mathring{A}$ ,  $^{\circ}$ ).

O1-C1	1.230(3)	N1-C1	1.340 (3)
O2-C6	1.231 (3)	N1-C5	1.435 (3)
O3-C7	1.334 (3)	N1-C2	1.462 (3)
O3-C3	1.452 (3)	N2-C6	1.312 (4)
O4-C7	1.196 (3)		
C7-O3-C3	117.66 (18)	C1-N1-C2	113.1 (2)
C1-N1-C5	122.7 (2)	C5 - N1 - C2	123.8 (2)

**Table 2** Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
$ \begin{array}{c} N2-H2A\cdotsO2^{i} \\ N2-H2B\cdotsO1^{ii} \end{array} $	0.85 (3)	2.14 (3)	2.974 (3)	166 (3)
	0.94 (4)	2.04 (4)	2.964 (4)	168 (3)

Symmetry codes: (i) -x + 1, -y, -z + 1; (ii) -x + 1, -y + 1, -z + 1.

C-bound H atoms were positioned geometrically and refined using the riding-model approximation, with C–H = 0.96–0.97 Å and  $U_{\rm iso}({\rm H})$  = 1.2 $U_{\rm eq}({\rm C})$ . H atoms attached to N atoms were located in a difference Fourier map and refined freely.

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

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